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# Selective extraction of intracellular components from the microalga *Chlorella vulgaris* by combined Pulsed Electric Field-Temperature treatment

P.R. Postma<sup>a</sup>, G. Pataro<sup>b,\*</sup>, M. Capitoli<sup>c</sup>, M.J. Barbosa<sup>d</sup>, R.H. Wijffels<sup>a,e</sup>, M.H.M. Eppink<sup>a</sup>,  
G. Olivieri<sup>a</sup>, G. Ferrari<sup>b,c</sup>

<sup>a</sup>: Bioprocess Engineering, AlgaePARC, Wageningen University, PO Box 16, 6700 AA  
Wageningen, The Netherland

<sup>b</sup>: Department of Industrial Engineering, University of Salerno, via Giovanni Paolo II  
132, Fisciano, SA, Italy

<sup>c</sup>: ProdAl Scarl - University of Salerno, via Ponte don Melillo, 84084 Fisciano (SA), Italy

<sup>d</sup>: Food & Biobased Reseach, AlgaePARC, Wageningen University and Research Centre,  
PO Box 17, 6700 AA Wageningen, The Netherlands

<sup>e</sup>: University of Nordland, Faculty of Biosciences and Aquaculture, N-8049 Bodø,  
Norway

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\* Corresponding author: Tel.: +39 089 96-4136/9439.

E mail addresses: [gpataro@unisa.it](mailto:gpataro@unisa.it) (G. Pataro), [richard.postma@wur.nl](mailto:richard.postma@wur.nl) (P.R. Postma)

**Abstract**

The synergistic effect of temperature (25-65 °C) and total specific energy input (0.55-1.11 kWh kg<sub>DW</sub><sup>-1</sup>) by pulsed electric field (PEF) on the release of intracellular components from the microalgae *Chlorella vulgaris* was studied. The combination of PEF with temperatures from 25-55 °C resulted in a conductivity increase of 75% as a result of cell membrane permeabilization. In this range of temperatures, 25-39% carbohydrates and 3-5% proteins release occurred and only for carbohydrate release a synergistic effect was observed at 55 °C. Above 55 °C spontaneous cell lysis occurred without PEF. Combined PEF-temperature treatment does not sufficiently disintegrate the algal cells to release both carbohydrates and proteins at yields comparable to the benchmark bead milling (40-45% protein, 48-58% carbohydrates).

**Keywords**

Microalgae; pulsed electric field; temperature; bead mill; mild cascade biorefinery

## 1 Introduction

Microalgae are promising for the production of multiple components like proteins, carbohydrates, lipids and pigments to be used as functional additives for cosmetic, nutraceutical, chemical, food and feed products as well as for the production of biofuels (Batista et al., 2013; Vanthoor-Koopmans et al., 2013). In order to have an economically feasible production process all intracellular components have to be used in a biorefinery approach (Günerken et al., 2015; Wijffels et al., 2010).

Biorefinery comprises the downstream processing (i.e. recovery, fractionation and purification) of added value ingredients from biomass. Most interesting products from microalgae are commonly stored either in the cytoplasm or in internal organelles. The complex cell structure of microalgae, comprises several organelles such as the chloroplast, mitochondria, Golgi apparatus, nucleus etc., and all these organelles have a different composition and structure (Eppink et al., 2012). In a cascade biorefinery approach, the first step after harvesting is the use of a cell disintegration technique able to break the cell wall and cell membranes facilitating the release of these high value added components from the cytoplasm and the internal organelles (Günerken et al., 2015; Vanthoor-Koopmans et al., 2013). However, cell disintegration should be done avoiding the use of severe processing conditions that could negatively affect the quality and purity of the extracts, diminishing the product value. Conventional cell disintegration techniques may cause complete cell disruption, thus fostering the non-selective release of all cell components. This will reduce the quality and purity of the extracts complicating the subsequent fractionation phase. Moreover, these techniques

could be very energy intensive. For instance bead milling is known to be a high-energy-consuming-technique requiring specific energy levels of 0.81, 1.71 or 7.64 kWh kg<sub>DW</sub><sup>-1</sup> for a biomass concentration of 145, 87.5 and 25 kg m<sup>-3</sup>, respectively (Postma et al., 2015). Therefore, there is a growing interest in finding mild cell disintegration methods, but effective enough to facilitate the release of the target compounds from the inner parts of the cells with low specific energy consumption.

In the past decade, pulsed electric field (PEF) has been claimed to be a promising mild technique able to induce the permeabilization of the microalgal cell membranes by electroporation and to enhance the spontaneous release of intracellular components (Toepfl et al., 2006). Recently, Goettel et al. (2013) investigated for the first time the use of PEF for the release of multiple intracellular components (protein, carbohydrates and lipids) from microalgae. Later on, other researchers (Coustets et al., 2014; Grimi et al., 2014; Lai et al., 2014; Luengo et al., 2015; Parniakov et al., 2015) have studied the effect of PEF on the recovery of single cell products, where a total specific energy consumption ranging between 0.4 –30.9 kWh kg<sub>DW</sub><sup>-1</sup> was applied. Originally PEF was widely investigated and applied in the medical field for electrochemotherapy (Jaroszeski et al., 2000) and gene transfer (i.e. electrotransformation) of microorganisms and plant cells (Kandušer and Miklavčič, 2009). Additionally, it has been applied in the food industry for either microbial inactivation (cold-pasteurization) or on mass transfer of liquids and valuable compounds from the inner parts of plant cells (extraction, drying) (Álvarez et al., 2006; Donsi et al., 2010; Kotnik et al., 2015; Pataro et al., 2011; Vorobiev and Lebovka, 2010). Moreover, Ganeva et al. (2003) showed that PEF can be a mild technique to

disintegrate yeast, where 70-90% of the total enzyme activity was maintained and large proteins up to 250 kDa could be released with a yield up to 50%. To our knowledge, these yields of large proteins have so far not been obtained for microalgae.

Several parameters influence the PEF efficacy, which mainly include electric field strength and total specific energy input (Pataro et al., 2014). In general, depending on the settings of these parameters reversible or irreversible pores are formed (Kotnik et al., 2015). However, in some cases, it is necessary to apply intense process conditions (high field strengths and energy inputs) to obtain a sufficient high permeabilization degree of the cell membrane (Pataro et al., 2011). Therefore, in order to obtain the required permeabilization effect with less severe processing conditions, or to achieve higher efficacy at the same treatment conditions, PEF has also been applied in a hurdle approach. For example, additive or synergistic effects have been observed by combining PEF with moderate heating above 35 °C for microbial inactivation (Timmermans et al., 2014). As per the literature survey, only Luengo et al. (2015) described the effect of temperature (10-40 °C) on the release of the pigment lutein during PEF treatment of microalga *Chlorella vulgaris*. They found that, at 25 kV cm<sup>-1</sup>, increasing the temperature of the biomass from 10 °C to 20 °C increased the extraction yield by 35%, but an increase lower than 10% was observed by raising the treatment temperature from 30 to 40 °C. Therefore, further studies are necessary to better elucidate the interactions between electric field and temperature and the dependence of the extraction yield of the target components on the process parameters of the combined treatment. To our knowledge this is the first attempt to

investigate the effect of a combined PEF-Temperature treatment on the selective release of multiple water soluble components from microalgae within a microalgae biorefinery concept.

The objective of this work was to investigate the effect of the processing temperature during PEF treatment of the microalga *Chlorella vulgaris* on the selective release of intracellular components. Release of ions (conductivity), carbohydrates and proteins was followed. Additionally, the effect of this treatment on the product quality was measured with gel electrophoresis and a Rubisco activity assay.

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## 2 Materials and Methods

### 2.1 Microalgae, cultivation and logistics

*C. vulgaris* (SAG 211-11b) was obtained from the Culture Collection of Algae at Göttingen University and was cultivated in M-8a medium as described by Kliphuis et al. (2010) using a 12L stirred tank photobioreactor (Postma et al., 2015). After harvesting, the biomass was concentrated by means of centrifugation (4000 x g, 15min and 4 °C) up to a final concentration ( $C_x$ ) of 25 kg<sub>DW</sub> m<sup>-3</sup> with an initial conductivity ( $\sigma$ ) of about 0.6 mS cm<sup>-1</sup> at 25 °C (Conductivity meter HI 9033, Hanna Instrument, Milan, Italy).

The concentrated biomass was pre-packed in high-density polyethylene bottles (Nalgene) and cooled to 4 °C. The transport of the biomass to ProDAI Scarl (University of Salerno, Fisciano (SA), Italy) was conducted by courier within 24 hours in an EPS box in which the refrigerated temperature was maintained using gel-packs. PEF treatments were performed on the delivery day. A sample of the concentrated biomass was taken and analyzed at Wageningen University on the shipment day as well as the delivery day. Results showed no influence of the microalgae transport to Salerno on the cell components of interest (proteins and carbohydrates) for this work (data not shown).

### 2.2 PEF experimental set-up

PEF experiments were conducted in a bench-scale continuous flow PEF system of which a schematic overview is shown in Figure 1. The PEF unit design was based on the unit described by Pataro et al. (2014), but adjusted in order to include two PEF

treatment zones. In short, a peristaltic pump (Pump Drive PD5201, Heidolph Instruments GmbH, Germany) provided a continuous flow  $Q$  of  $33 \text{ mL min}^{-1}$  of the algae biomass suspension. Prior to entering each of the two PEF treatment zones, the algae suspension flowed through a stainless steel coil immersed in a water heating bath to control the temperature between  $25 - 65 \text{ }^\circ\text{C}$ . Each PEF treatment zone consisted of a module made of two co-linear cylindrical treatment chambers, hydraulically connected in series, with an inner radius ( $r$ ) of  $1.5 \text{ mm}$  and a gap distance ( $L$ ) of  $4 \text{ mm}$ . The geometry of each treatment chamber was previously described by (Pataro et al., 2012) and is schematically shown in Figure 1B.

Monopolar square wave electric pulses were supplied by a high voltage pulse generator (Diversified Technology Inc., Bedford, WA, USA). The voltage and current signals at the treatment chamber were measured by a high voltage probe (P6015A, Tektronix, Wilsonville, OR, USA) and a Rogowsky coil (2-0.1W, Stangenes, Inc., USA), respectively. The measurements were recorded and displayed using a  $300 \text{ MHz}$  digital oscilloscope (TDS 3034B, Tektronix, Wilsonville, OR, USA) connected to a PC. An example of typical pulse waveforms at the treatment chamber is shown in Figure 2. The theoretical electric field intensity ( $E$ ) was evaluated as the applied voltage ( $U$ ) divided by the gap between electrodes ( $L$ ). However, it is known that the actual electric field strength in the PEF treatment zone of a co-linear chamber is often lower than the theoretical value ( $E$ ) and suffers from a non-uniform distribution and intensity peaks near the insulator edges (Buckow et al., 2011). Therefore, in this work, the actual average electric field strength  $E_{av}$  ( $\text{kVcm}^{-1}$ ) applied in the treatment zone was

evaluated according to the following equation previously proposed by Buckow et al.

(2011):

$$E_p \cdot E_{av}^{-1} = 1 + \alpha \cdot r \cdot L^{-1} \quad (1)$$

where  $E_p$  (kV cm<sup>-1</sup>) is the theoretical peak electric field strength evaluated as the peak voltage divided by the inter-electrode gap, and  $\alpha$  is a coefficient which assumes a value of 0.4475 for an inner radius of 1.5 mm (Pataro et al., 2012). The total specific energy input  $W_{PEF}$  (kWh kg<sub>DW</sub><sup>-1</sup>) was calculated as a function of the total number of pulses  $n$  applied in the four treatment chambers, the volume  $v_i$  (mL) of each treatment zone and the biomass concentration  $C_x$  (kg<sub>DW</sub> m<sup>-3</sup>).  $U(t)$  and  $I(t)$  represent the voltage across the electrodes and the current intensity through the product at time  $t$  (s), respectively:

$$W_{PEF} = \frac{n}{3.6 \cdot 10^{-3} \cdot v_i \cdot C_x} \int_0^{\infty} U(t) \cdot I(t) dt \quad (2)$$

$$\text{in which: } n = \frac{v_i \cdot N_{TC}}{\Phi} \cdot f \quad (3)$$

Where,  $N_{TC}$  is the total number of treatment chambers,  $\Phi$  is the flow rate (mL s<sup>-1</sup>) and  $f$  is the pulse repetition frequency (Hz). During the treatment, a single voltage value of 8 kV was set resulting in a theoretical peak electric field strength ( $E_p$ ) of 20 kV cm<sup>-1</sup> which corresponds, according to Eq. (1), to an average electric field ( $E_{av}$ ) of 17.1 kV cm<sup>-1</sup>.

The pulse length was fixed to 5  $\mu$ s and the pulse frequency was adjusted between 50 – 200 Hz to provide a specific energy input ( $W_{PEF}$ ) of 0.55 and 1.11 kWh kg<sup>-1</sup> dry weight (kWh kg<sub>DW</sub><sup>-1</sup>) at each processing temperature investigated (25-65°C). Four

thermocouples were used to measure the product temperature at the inlet and outlet of the PEF chamber module.

An overview of the conducted experiments and process conditions is shown in Table 1. At the exit of the unit the treated algae suspension was collected in plastic tubes and placed in an ice water bath. After cooling, the samples were allowed to stand for 1 hour at 25 °C under shaking at 140 rpm to allow intracellular components to diffuse out of the cells. After this resting time, the cell suspensions were centrifuged (10 min, 5300 x g) and the supernatant was transferred to fresh tubes and stored at -20 °C until further analysis.

### 2.3 Synergy of PEF and temperature

Energy was provided via two ways in the current study: (1) heating (increased processing temperature using water heating baths) and (2) via electrical pulses, abbreviated as  $W_T$  and  $W_{PEF}$ , respectively. For the algae suspension (97.5% w/w water, 2.5% w/w algae) a specific heat capacity of  $1.16 \cdot 10^{-3} \text{ kWh (kg K)}^{-1}$  was used assuming a suspension of only water. When increasing the temperature from 25 up to 65 °C, up to an additional  $1.83 \text{ kWh kg}_{\text{DW}}^{-1}$  energy is consumed. Where  $W_{PEF}$  was 0.55 or 1.11 kWh  $\text{kg}_{\text{DW}}^{-1}$ , on the laboratory scale plant used in this work, this had a substantial effect (i.e.  $W_T \approx W_{PEF}$ ) on the total amount of consumed energy for a single operating condition. Nevertheless, on an industrial scale, heat exchangers could be used to recover the energy from the outlet stream. Besides, the energy input of the PEF ( $W_{PEF}$ ) also increased the temperature of the suspension (5-10 °C). This increase can be sufficient to exchange enough energy between the outlet and inlet, reducing the amount

additional  $W_T$  to a minimum. Therefore, only  $W_{PEF}$  was considered in comparison to other studies.

## 2.4 Bead mill experimental procedure

The bead mill (Dyno Mill Research Lab, WAB AG Maschinenfabrik, Muttenz, Switzerland) operation procedure was previously described (Postma et al., 2015). In short, 1 mm  $ZrO_2$  beads (Tosoh YTZ<sup>®</sup>) were used at a filling percentage of 65% v/v. The bead mill was operated with an agitator speed of 6, 9 or 12  $m\ s^{-1}$  and a biomass concentration of 25  $kg_{DW}\ m^{-3}$  was used. The specific energy consumption for bead milling is expressed as  $W_{BM}$  ( $kWh\ kg_{DW}^{-1}$ ).

## 2.5 Analytical methods

### 2.5.1 Protein analysis

The total protein content on biomass dry weight (DW) and the water soluble protein content of supernatants before and after PEF treatment were determined according to Postma et al. (2015). In short, for total protein content on DW, 6 mg of freeze dried algae were bead beaten in 1.0 mL lysis buffer I (60 mM Tris, 2% SDS, pH 9.0) in a lysing matrix E tube (6914-500, MP Biomedicals Europe, France). The tubes were beaten using a bead beater (Precellys 24, Bertin Technologies, France) for 3 cycles of 60 s at 6500 RPM with 120 s breaks between cycles.

For analysis of the water soluble protein content, supernatant obtained from PEF-treated samples was diluted 2 times using lysis buffer II (120 mM Tris, 4% SDS, pH 9.0).

Subsequently, samples for both total protein content on DW and water soluble protein content from supernatant were incubated at 100 °C for 30 min before quantification using a commercial kit (DC™ Protein assay, Bio-Rad, USA) similar to the Lowry assay (Lowry et al., 1951). Bovine serum albumin (A7030, Sigma-Aldrich, USA) was used as protein standard. The absorbance was measured at 750 nm. The protein yield ( $Y_p$ ) was expressed as:

$$Y_p = \frac{C_{p,sup}}{C_{p,biomass}} \quad (4)$$

Where  $C_{p,sup}$  is the protein content in the supernatant (%<sub>DW</sub>) and  $C_{p,biomass}$  is the total protein content on DW (%<sub>DW</sub>).

### 2.5.2 Carbohydrate analysis

In order to analyze the carbohydrate content on biomass DW, ~1 mg of algae DW was hydrolyzed in 1 mL 2.5 M HCl in a heat block at 100 °C for 3 hours. Samples were neutralized using 1 mL of 2.5 M NaOH.

Both, hydrolyzed samples and supernatant of PEF treated samples were analyzed according to DuBois et al. (1956). 0.2 mL of 5% w/w phenol and 1 mL of concentrated sulfuric acid were added to 0.2 mL of (diluted) sample (hydrolyzed sample or supernatant obtained from PEF). The samples were incubated at 35 °C for 30 minutes before reading of the absorbance at 485 nm against a blank of 0.2 mL 5% w/w phenol, 1 mL concentrated sulfuric acid and 0.2 mL of deionized water. Glucose was used as a standard. The carbohydrate yield ( $Y_c$ ) was expressed as:

$$Y_c = \frac{C_{c,sup}}{C_{c,biomass}} \quad (5)$$

In which  $C_{c,sup}$  is the carbohydrate content in the supernatant (%<sub>DW</sub>) and  $C_{c,biomass}$  is the total carbohydrate content on DW (%<sub>DW</sub>).

### 2.5.3 Polyacrylamide gel electrophoresis

Supernatant samples from PEF experiments were thawed and 6x concentrated using Amicon Ultra-0.5 3K (Merck Millipore, USA) centrifugal tubes. Concentrated samples were kept on ice until further use.

Native PAGE was conducted using a 4-20% Criterion TGX gel (#567-1094, Biorad). 50 µl of Native sample buffer (#161-0738, Biorad) and 125 µg of protein was mixed and made up to 100 µL using Milli-Q®. 25 µg of protein was loaded per lane. NativeMark™ (LC0725, life technologies) was used as marker for size estimation. Tris/Glycine (#161-0734, Biorad) was used as running buffer at 200 V constant for 35 min.

Bio-Safe Coomassie stain (#161-0787, Biorad) was used to stain the Native PAGE gel for 120 min followed by overnight rinsing with de-ionized water to increase background contrast before scanning.

### 2.5.4 Rubisco activity assay

The Rubisco activity was analyzed spectrophotometrically according to Desai et al. (2014). NADH oxidation was measured at 340 nm using quartz cuvettes over a period of 6 minutes. The Rubisco activity was calculated using a molar extinction

coefficient of  $6.22 \text{ mM}^{-1}$ . The final reaction mixture (3 mL) contains 259 mM Tris, 5 mM magnesium chloride, 67 mM potassium bicarbonate, 0.2 mM  $\beta$ -nicotinamide adenine dinucleotide (reduced form), 5 mM adenosine 5' -triphosphate, 5 mM glutathione (reduced form), 0.5 mM D-ribulose 1,5-di-phosphate, 5 units alpha-glycerophosphate dehydrogenase trios phosphate isomerase, and 5 units glyceraldehyde-3-phosphate dehydrogenase/3-phosphoglyceric phosphokinase. Either, PEF treated supernatant or bead mill supernatant was added just before the measurement, a reaction mixture without Rubisco was used as blank.

### 3 Results and Discussion

The biomass composition of *C. vulgaris* used in this study was quantified as follows: 61.1 %<sub>DW</sub> protein and 16.2 %<sub>DW</sub> carbohydrates. The results of this study will be presented in terms of extraction yields expressed with respect to this composition according to Eq. (4) and (5). Cell membrane permeabilization was monitored by measurement of electrical conductivity (Donsì et al., 2010). The release of relatively small and large molecules was monitored by the release of carbohydrates and proteins, respectively. Furthermore, a detailed analysis on the size distribution and activity of the released protein is presented in comparison to our benchmark bead milling. Subsequently, the role of PEF in a biorefinery approach is discussed.

### 3.1 Effect of combined PEF-temperature treatment on the release of macromolecules

#### 3.1.1 Permeabilization of the algal cells

In order to quantify the maximal increase in conductivity after combined PEF-temperature treatment, a sample of the microalgal suspension was subjected to a beat beater step followed by measurement of the conductivity. The microalgal suspension had a maximal absolute increase of  $1.0 \text{ mS cm}^{-1}$  with respect to the fresh sample ( $0.6 \text{ mS cm}^{-1}$ ). This maximal conductivity increase was further used as a benchmark value to calculate the relative increase of conductivity for PEF treated samples 1 h after PEF treatment. From the results shown in Figure 3A it can be observed that there was a strong increase in conductivity after PEF treatment under all conditions applied. This is a clear indication that PEF caused permeabilization of the algal cell resulting in leakage of small ions as also observed by Goettel et al. (2013) and Grimi et al. (2014). Besides, it can be observed that processing temperatures between 25 and 55 °C did not result in additional spontaneous release of ions without PEF treatment. However, when the algae were heated up to 65 °C without PEF treatment, a strong increase of the conductivity up to 62.5% of the maximal possible was observed. As a result of the PEF treatment at 25 – 55 °C, the conductivity increased up to 60-75% of the maximal conductivity value. Applying PEF at an energy input of 0.55 or  $1.11 \text{ kWh kg}_{\text{DW}}^{-1}$  at 65 °C only resulted in a further increase of the conductivity up to 65.5% or 67.5% respectively. The maximal relative conductivity increase was 75% achieved after a PEF treatment at  $1.11 \text{ kWh kg}_{\text{DW}}^{-1}$  at 45 °C. From these results, which are in agreement

with the findings of Grimi et al. (2014), it can be concluded that, under the processing conditions investigated, no complete extraction of ionic components was achieved.

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### 3.1.2 Release of carbohydrates

The analysis of the carbohydrates in the supernatant (Figure 3B) shows that only a small fraction ( $Y_c$ : < 5% of biomass carbohydrate content) of the total carbohydrates content in the microalgae was released at processing temperatures between 25 and 55 °C without any PEF treatment. However, when the temperature was further increased up to 65 °C, a substantial increase in the release of carbohydrates was observed. This was likely due to the thermal disintegration of the cell membranes, resulting in an extraction yield of more than 35% of the total carbohydrate content. Figure 3B also shows the carbohydrate release yields achieved after the combined PEF-temperature treatments. When PEF was applied at processing temperatures between 25 and 45 °C, carbohydrate yields between 22 and 25% were obtained. Nevertheless, no positive interaction between PEF and temperature can be noted in this range of temperatures. Further increase of the processing temperature up to 55°C, instead, showed a clear synergistic effect of the combined treatment leading to an increase of the carbohydrate yield up to 39%. However, no difference could be detected at any temperature when the total specific energy input was increased from 0.55 to 1.11 kWh kg<sub>DW</sub><sup>-1</sup>. This synergistic effect was most likely caused by a less stable cell membrane due to the increased temperature, making the lipid bilayer of the cell membrane more sensitive for electric pulses (Timmermans et al., 2014) allowing a larger amount of carbohydrates to be released. This effect could even be further enhanced because at increased temperature values, diffusivity and solubility of carbohydrates tend to be higher. In contrast, no further release of carbohydrates was observed at a processing temperature of 65 °C when PEF was

applied. This is likely due to the fact that, at this high temperature value, the thermal effect was enough to break the cell membranes masking the PEF effect. Goettel et al. (2013) reported a carbohydrate release of about  $8 \text{ g L}^{-1}$  using a specific energy input of  $0.40 \text{ kWh kg}_{\text{DW}}^{-1}$  from a suspension of  $109 \text{ g}_{\text{DW}} \text{ kg}^{-1}$  with the microalgae *Auxenochlorella protothecoides*. Assuming a carbohydrate content of 33% on DW (Bohutskyi et al., 2015), a carbohydrate yield of 22% was achieved which is in the same range as the current study.

### 3.1.3 Release of water soluble protein

Figure 3C shows the protein yield achieved after either heating or combined PEF-temperature treatment. The results show that no or hardly any protein could be detected in the supernatant of the suspension when no PEF treatment was applied at a processing temperature between 25 and 55 °C. This is either because no protein was released or the protein content was below the limit of quantitation ( $0.05 \text{ mg mL}^{-1}$ ). However, similarly to the results on conductivity (Figure 3A) and release of carbohydrates (Figure 3B), a substantial increase in the released protein up to an extraction yield of 3.7% was found at a processing temperature of 65 °C without any PEF treatment. The application of a PEF treatment at room temperature resulted in an extraction yield of 3.2% at  $0.55 \text{ kWh kg}_{\text{DW}}^{-1}$  and 3.6% at  $1.11 \text{ kWh kg}_{\text{DW}}^{-1}$ . A slight synergistic effect was observed when the electrical treatment was combined with heating of the biomass. However, the absolute maximum yields are still a tenfold lower than the yields obtained with the bead mill ( $Y_p$ : 40-45% of biomass protein content at  $25 \text{ kg}_{\text{DW}} \text{ m}^{-3}$ ) (Postma et al., 2015). Nevertheless, it appeared that above a

certain critical temperature, no further improvement could be observed. For the PEF treatment at 0.55 and 1.11 kWh kg<sub>DW</sub><sup>-1</sup>, the maximum protein extraction was obtained at 55 °C ( $Y_p$ : 4%) and 45 °C ( $Y_p$ : 4.4%), respectively.

Although for mild treatments generally temperatures below 35 °C are used to prevent any damage to the protein structure, this is only a prerequisite if long treatment times (i.e. order of minutes to hours) are applied. In the current work, the residence time of the biomass inside the treatment chambers was only 0.22 s, while the total residence time at each processing temperature was lower than 25 s. Moreover, the treated algae suspension collected at the exit of the PEF unit was immediately cooled in a water-ice bath. Studies on the denaturation kinetics of whey proteins showed that it takes over 200 seconds at 65 °C to denature 1% of  $\beta$ -lactoglobulins or  $\alpha$ -lactalbumin (Dannenberg and Kessler, 1988).

In recent PEF studies, similar protein yields and corresponding specific energy inputs ( $W'_{PEF}$ ) to the results presented in this study were reported. Grimi et al. (2014) found a relative yield of 3.6% with an energy consumption of 1.50 kWh kg<sub>DW</sub><sup>-1</sup> using *Nannochloropsis sp.* Parniakov et al. (2015) even obtained a protein yield between 5-10% although this required 4.00 kWh kg<sub>DW</sub><sup>-1</sup>. In addition, these authors used frozen/thawed *Nannochloropsis sp.* algae prior to the application of PEF, which most likely weakened/damaged the cell structure. On the other hand, Goettel et al. (2013) were able to enhance the spontaneous release of protein slightly (from 8  $\mu$ g L<sup>-1</sup> before PEF to 10.5  $\mu$ g L<sup>-1</sup> after PEF) from fresh cells of *A. protothecoides* at a specific energy input of 0.40 kWh kg<sub>DW</sub><sup>-1</sup>.

### 3.2 Protein size distribution and activity

Since low protein yields ( $Y_p$ : < 5 % of biomass protein content) were obtained in comparison to the yields obtained with the bead mill ( $Y_p$ : 40 - 45% of biomass protein content at a  $C_x$  of 25 kg<sub>DW</sub> m<sup>-3</sup>) (Postma et al., 2015), and a substantial amount of algal protein is water soluble (Safi et al., 2014), it is expected that only small proteins were released. Therefore, the size distribution of the released proteins was determined via gel electrophoresis. This gives better understanding on the release behavior and location of the proteins.

#### 3.2.1 Native PAGE

Native PAGE provides understanding whether the released protein is negatively affected in size (i.e. degradation or aggregation). Figure 4 shows the Native PAGE gel in which bead mill samples (lanes 14-16) are compared to the PEF treated samples at an energy input of 1.11 kWh kg<sub>DW</sub><sup>-1</sup> at 25- 65 °C (lane 2-10, 12, 13). From lanes 2-13 it can be observed that a wide range of proteins of different molecular sizes was released by combined PEF-temperature treatment. From 45 °C onwards, a more intense group of proteins is visible between 20-66 kDa, and at 65°C even additional bands occur. This indicates that elevated processing temperatures do negatively affect the native state of the released protein, irreversibly damaging the protein structures, despite the short processing time as described in section 3.1.3. Conclusively, only processing temperatures up to 35 °C should be applied if native proteins are desired. Comparing the overall profile of the proteins released by bead milling and PEF, it can be observed that the bead mill samples reveal a strong band at ~540 kDa (i.e. the size of native

Rubisco) next to a large range of proteins in different sizes, both larger and smaller. The samples subjected to PEF also reveal a band at ~540 kDa although less distinct. Based on the soluble protein (Lowry) assay, equal amounts of protein were loaded on the gel per lane. However, it can be observed that the higher molecular weight proteins are more distinct in the bead mill samples than in the PEF samples. Instead, below 20 kDa an intense band of low molecular weight protein material can be observed which is more pronounced for the PEF samples. To summarize, according to the results, it appears that PEF releases more small proteins rather than large proteins.

Similar to the observations in this work, Azencott et al. (2007) found that Bovine Serum Albumin (BSA) was able to move across the cell wall and cell membrane of a wild-type *Chlamydomonas reinhardtii* when subjected to electrical pulses. Although, virtually all cells were able to take up the fluorescent dye calcein (~0.6 kDa), only a fraction of the much larger BSA (~66 kDa) was taken up.

### 3.2.2 Rubisco activity

To confirm whether the released multimeric Rubisco was still biologically active a Rubisco activity assay was performed. Figure 5 shows the specific Rubisco activity of the protein released in the supernatant after PEF treatment (processing temperature: 35 °C,  $W_{PEF}$ : 1.11 kWh kg<sub>DW</sub><sup>-1</sup>) in comparison to a supernatant sample obtained from bead milling (agitator speed: 9 m s<sup>-1</sup>,  $W_{BM}$ : 14.1 kWh kg<sub>DW</sub><sup>-1</sup>). It can be observed that the specific activity of the bead mill sample was about 10 times higher than the PEF sample, indicating that the purity (i.e. amount of Rubisco per amount of total protein) of the PEF sample was lower. This confirms the observations from the Native PAGE gel

leading to the suggestion that during PEF treatment less intracellular organelles were disintegrated than during bead milling. Bead milling completely disintegrates the algal cells including larger and smaller internal organelles. Therefore, also Rubisco stored in the chloroplast (free in stroma and/or inside pyrenoids) would be released. Nevertheless, since active Rubisco was observed after combined PEF-temperature treatment, it is likely that part of the chloroplast was disintegrated causing free Rubisco from the stroma to diffuse out of the cell. Though, this hypothesis cannot be confirmed based on the current results.

### 3.3 The role of PEF in a cascade biorefinery

A multi-stage biorefinery should exploit the full potential of the cell in terms of compartmentalization and products and, at the same time, require a low energy input (Eppink et al., 2012). Mahnič-Kalamiza et al. (2014) envisioned the application of PEF as a first disintegration step in such a multi-stage biorefinery followed by two extraction stages. In the first stage, water soluble components should be extracted and in a second stage an environmentally friendly solvent could be applied to extract pigments or other hydrophobic components. In addition, Coons et al. (2014) described that no more than 10% of the total available energy from the produced algae  $\bar{W}_{Algae}$  (6.82 kWh  $\text{kg}_{\text{DW}}^{-1}$ ) should be utilized for the disintegration, extraction and fractionation, abbreviated as  $\bar{W}_{Biorefinery}$ .

Our findings indicate that maximum 39% of the carbohydrates and less than 5% of the proteins could be released, but only at a combined PEF-temperature treatment at 55 °C and 0.55 or 1.11 kWh  $\text{kg}_{\text{DW}}^{-1}$ . Even though, at a specific energy input of 0.55

$\text{kWh kg}_{\text{DW}}^{-1}$  the energy target  $\bar{W}_{\text{Biorefinery}}$  of  $0.682 \text{ kWh kg}_{\text{DW}}^{-1}$  is exceeded, the results of the native PAGE showed that from  $45 \text{ }^\circ\text{C}$  onwards the proteins seem to be negatively affected. Albeit a lower processing temperature of  $35 \text{ }^\circ\text{C}$  yields active Rubisco, a carbohydrate and protein yield of only 25% and 3.8% were obtained, respectively.

In the ideal case, PEF would allow selective release of small soluble molecules (e.g. carbohydrates) resulting in relative pure fractions without negatively harming the other components. This requires the optimization of the PEF processing conditions in order to further increase the extraction yield of these small molecules. To this purpose, it should be taken into account that our results were obtained by combining moderate temperatures with a PEF treatment applied at a relatively low theoretical field strength value ( $20 \text{ kV/cm}$ ). In addition, this value was further reduced by 15% due to the intrinsic non-uniform distribution of the electric field inside the treatment zone of the co-linear chamber used in this work, so that the actual average electric field strength applied was  $17.1 \text{ kV cm}^{-1}$ . Therefore, further experimental work should be carried in order to test whether, for example, the combination of moderate temperature with higher field strength as well as improvements of the treatment uniformity by modifications of the chamber geometry (e.g. insulator and electrode ring shape) may result in a higher extraction yield of small molecules like carbohydrates, without negatively affecting the other components.

In this way extensive fractionation of complex protein and carbohydrate mixtures (e.g. by membrane filtration) could be omitted while reducing  $\bar{W}_{\text{Biorefinery}}$ . After the PEF stage, effort should be made in order to recover the high-value water

soluble proteins and likewise to allow the recovery of the remaining carbohydrates in a subsequent stage before any solvent is applied (Figure 6). Bead milling has shown to be a promising candidate to release up to 45% of the water soluble protein (Postma et al., 2015) and 48-58% of carbohydrates (data not shown).

To conclude, further studies are necessary in order to optimize the processing conditions of the combined PEF-temperature treatment as well as its integration in a cascade biorefinery in order to maximize the selective recovery of the high value components with minimal energy consumption.

#### **4 Conclusion**

PEF showed to be an effective technique to release small ionic solutes and carbohydrates up to 75% and 39%, respectively. Nevertheless, permeabilization was not effective enough to release high quantities of large molecules such as protein (< 5%). Though, the released protein fraction contained the biologically active multimeric Rubisco showing that PEF is a mild technique at 35 °C. In conclusion, under the processing conditions investigated in this work, the combined PEF-temperature treatment seems not able to sufficiently disintegrate the algal cells to release both carbohydrates and proteins at yields comparably to the benchmark bead milling (40-45% protein, 48-58% carbohydrates).

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## References

- [1] Álvarez, I., Condón, S., Raso, J., 2006. Microbial Inactivation by Pulsed Electric Fields, in: Raso, J., Heinz, V. (Eds.), *Pulsed Electric Fields Technology for the Food Industry*, Food Engineering Series. Springer US, pp. 97–129.
- [2] Azencott, H.R., Peter, G.F., Prausnitz, M.R., 2007. Influence of the cell wall on intracellular delivery to algal cells by electroporation and sonication. *Ultrasound Med. Biol.* 33, 1805–1817.
- [3] Batista, A.P., Gouveia, L., Bandarra, N.M., Franco, J.M., Raymundo, A., 2013. Comparison of microalgal biomass profiles as novel functional ingredient for food products. *Algal Res.* 2, 164–173.
- [4] Bohutskyi, P., Ketter, B., Chow, S., Adams, K.J., Betenbaugh, M.J., Allnut, F.C.T., Bouwer, E.J., 2015. Anaerobic digestion of lipid-extracted *Auxenochlorella protothecoides* biomass for methane generation and nutrient recovery. *Bioresour. Technol.* 183, 229–239.
- [5] Buckow, R., Baumann, P., Schroeder, S., Knoerzer, K., 2011. Effect of dimensions and geometry of co-field and co-linear pulsed electric field treatment chambers on electric field strength and energy utilisation. *J. Food Eng.* 105, 545–556.
- [6] Coons, J.E., Kalb, D.M., Dale, T., Marrone, B.L., 2014. Getting to low-cost algal biofuels: A monograph on conventional and cutting-edge harvesting and extraction technologies. *Algal Res.* 6, Part B, 250–270.
- [7] Coustets, M., Joubert-Durigneux, V., Hérault, J., Schoefs, B., Blanckaert, V., Garnier, J.-P., Teissié, J., 2014. Optimization of proteins electroextraction from microalgae by a flow process. *Bioelectrochemistry*.
- [8] Dannenberg, F., Kessler, H.-G., 1988. Reaction Kinetics of the Denaturation of Whey Proteins in Milk. *J. Food Sci.* 53, 258–263.

- [9] Desai, R.K., Streefland, M., Wijffels, R.H., Eppink, M.H.M., 2014. Extraction and stability of selected proteins in ionic liquid based aqueous two phase systems. *Green Chem.*
- [10] Donsì, F., Ferrari, G., Pataro, G., 2010. Applications of Pulsed Electric Field Treatments for the Enhancement of Mass Transfer from Vegetable Tissue. *Food Eng. Rev.* 2, 109–130.
- [11] DuBois, M., Gilles, K.A., Hamilton, J.K., Rebers, P.A., Smith, F., 1956. Colorimetric Method for Determination of Sugars and Related Substances. *Anal. Chem.* 28, 350–356.
- [12] Eppink, M.H.M., Barbosa, M.J., Wijffels, R.H., 2012. Biorefining of microalgae: Production of high value products, bulk chemicals and biofuels, in: Posten, C., Walter, T., C. (Eds.), *Microalgal Biotechnology: Integration and Economy*. De Gruyter, Berlin.
- [13] Ganeva, V., Galutzov, B., Teissié, J., 2003. High yield electroextraction of proteins from yeast by a flow process. *Anal. Biochem.* 315, 77–84.
- [14] Goettel, M., Eing, C., Gusbeth, C., Straessner, R., Frey, W., 2013. Pulsed electric field assisted extraction of intracellular valuables from microalgae. *Algal Res.* 2, 401–408.
- [15] Grimi, N., Dubois, A., Marchal, L., Jubeau, S., Lebovka, N.I., Vorobiev, E., 2014. Selective extraction from microalgae *Nannochloropsis* sp. using different methods of cell disruption. *Bioresour. Technol.* 153, 254–259.
- [16] Günerken, E., D'Hondt, E., Eppink, M.H.M., Garcia-Gonzalez, L., Elst, K., Wijffels, R.H., 2015. Cell disruption for microalgae biorefineries. *Biotechnol. Adv.* 33, 243–260.
- [17] Jaroszeski, M.J., Heller, R., Gilbert, R., 2000. *Electrochemotherapy, Electrogenotherapy, and Transdermal Drug Delivery: Electrically Mediated Delivery of Molecules to Cells*. Springer Science & Business Media.
- [18] Kandušer, M., Miklavčič, D., 2009. Electroporation in Biological Cell and Tissue: An Overview, in: *Electrotechnologies for Extraction from Food Plants and Biomaterials*, Food Engineering Series. Springer New York, pp. 1–37.
- [19] Kliphuis, A.M.J., de Winter, L., Vejrazka, C., Martens, D.E., Janssen, M., Wijffels, R.H., 2010. Photosynthetic efficiency of *Chlorella sorokiniana* in a turbulently mixed short light-path photobioreactor. *Biotechnol. Prog.* 26, 687–696.
- [20] Kotnik, T., Frey, W., Sack, M., Haberl Meglič, S., Peterka, M., Miklavčič, D., 2015. Electroporation-based applications in biotechnology. *Trends Biotechnol.*

- [21] Lai, Y.S., Parameswaran, P., Li, A., Baez, M., Rittmann, B.E., 2014. Effects of pulsed electric field treatment on enhancing lipid recovery from the microalga, *Scenedesmus*. *Bioresour. Technol.* 173, 457–461.
- [22] Lowry, O.H., Rosebrough, N.J., Farr, A.L., Randall, R.J., 1951. Protein measurement with the folin phenol reagent. *J. Biol. Chem.* 193, 265–275.
- [23] Luengo, E., Martínez, J.M., Bordetas, A., Álvarez, I., Raso, J., 2015. Influence of the treatment medium temperature on lutein extraction assisted by pulsed electric fields from *Chlorella vulgaris*. *Innov. Food Sci. Emerg. Technol., Applications of PEF for food processing* 29, 15–22.
- [24] Mahnič-Kalamiza, S., Vorobiev, E., Miklavčič, D., 2014. Electroporation in Food Processing and Biorefinery. *J. Membr. Biol.* 247, 1279–1304.
- [25] Parniakov, O., Barba, F.J., Grimi, N., Marchal, L., Jubeau, S., Lebovka, N., Vorobiev, E., 2015. Pulsed electric field and pH assisted selective extraction of intracellular components from microalgae *Nannochloropsis*. *Algal Res.* 8, 128–134.
- [26] Pataro, G., De Lisi, M., Donsì, G., Ferrari, G., 2014. Microbial inactivation of *E. coli* cells by a combined PEF–HPCD treatment in a continuous flow system. *Innov. Food Sci. Emerg. Technol.* 22, 102–109.
- [27] Pataro, G., Lamberti, P., Ferrari, G., 2012. Mathematical modelling of the electric field distribution in a co-field chamber of a PEF system, in: *Proceeding of International Conference Bio & Food Electrotechnologies (BFE2012)*. Salerno, Italy, pp. 13–17.
- [28] Pataro, G., Senatore, B., Donsì, G., Ferrari, G., 2011. Effect of electric and flow parameters on PEF treatment efficiency. *J. Food Eng.* 105, 79–88.
- [29] Postma, P.R., Miron, T.L., Olivieri, G., Barbosa, M.J., Wijffels, R.H., Eppink, M.H.M., 2015. Mild disintegration of the green microalgae *Chlorella vulgaris* using bead milling. *Bioresour. Technol., Advances in biofuels and chemicals from algae* 184, 297–304.
- [30] Safi, C., Ursu, A.V., Laroche, C., Zebib, B., Merah, O., Pontalier, P.-Y., Vaca-Garcia, C., 2014. Aqueous extraction of proteins from microalgae: Effect of different cell disruption methods. *Algal Res.* 3, 61–65.
- [31] Timmermans, R.A.H., Nierop Groot, M.N., Nederhoff, A.L., van Boekel, M.A.J.S., Matser, A.M., Mastwijk, H.C., 2014. Pulsed electric field processing of different fruit juices: Impact of pH and temperature on inactivation of spoilage and pathogenic micro-organisms. *Int. J. Food Microbiol.* 173, 105–111.

- [32] Toepfl, S., Heinz, V., Knorr, D., 2006. Applications of Pulsed Electric Fields Technology for the Food Industry, in: Raso, J., Heinz, V. (Eds.), Pulsed Electric Fields Technology for the Food Industry, Food Engineering Series. Springer US, pp. 197–221.
- [33] Vanthoor-Koopmans, M., Wijffels, R.H., Barbosa, M.J., Eppink, M.H.M., 2013. Biorefinery of microalgae for food and fuel. *Bioresour. Technol.* 135, 142–149.
- [34] Vorobiev, E., Lebovka, N., 2010. Enhanced Extraction from Solid Foods and Biosuspensions by Pulsed Electrical Energy. *Food Eng. Rev.* 2, 95–108.
- [35] Wijffels, R.H., Barbosa, M.J., Eppink, M.H.M., 2010. Microalgae for the production of bulk chemicals and biofuels. *Biofuels Bioprod. Biorefining* 4, 287–295.

### Figure captions

**Figure 1 A)** Schematic overview of continuous flow PEF system. O: oscilloscope, UB: untreated biomass, ST: magnetic stirrer, HVPG: high voltage pulse generator, P: peristaltic pump, WB: water bath, HV+: high voltage, T: thermocouple, TC: treatment chamber, TB: treated biomass, WIB: water ice bath; and **B)** Dimensions and geometry of a single co-linear PEF treatment chamber in axis-symmetrical configuration. GR: ground electrode; HV: High Voltage electrode; L: gap distance (4 mm); r: inner radius (1.5 mm) (adapted from Pataro et al., 2012).

Figure 2 Typical pulse waveforms captured at the treatment chamber. A) Voltage  $U$  (blue) and Electric field strength  $E$  (red) and B) current  $I$  (green).  $E=U(t)/L$ .

**Figure 3 A)** Relative increase of electrical conductivity of the biomass suspension evaluated with respect to the conductivity of the bead beaten sample ( $1.6 \text{ mS cm}^{-1}$ ), **B)** Carbohydrate yield, and **C)** Protein yield as a function of the total specific energy input

$W_{PEF}$  ( $E_{ov}=17.1 \text{ kV cm}^{-1}$ ) and for different processing temperatures. The conductivity increase, carbohydrate release and protein release were measured 1h after PEF treatment.

**Figure 4** Native PAGE gel for PEF samples treated at  $W_{PEF} 1.11 \text{ kWh kgDW}^{-1}$ . Values in kDa. M: marker; 2: 25 °C; 3: 25 °C; 4: 35 °C; 5: 35 °C; 6: 45 °C; 7: 45 °C; 8: 45 °C; 9: 55 °C; 10: 55 °C; 11: 65 °C (no PEF); 12: 65 °C; 13: 65 °C; 14: bead mill  $6 \text{ m s}^{-1}$ ; 15: bead mill  $9 \text{ m s}^{-1}$ ; bead mill  $12 \text{ m s}^{-1}$

**Figure 5** Specific Rubisco activity (Units/mg protein) of supernatant obtained from PEF ( $17.1 \text{ kV cm}^{-1}$ ,  $1.11 \text{ kWh kgDW}^{-1}$ , and  $35 \text{ °C}$ ), and from bead milling (BM) ( $9 \text{ m s}^{-1}$ ).

Figure 6 A schematic representation of a single disintegration approach (top) compared to a multi-stage or cascade biorefinery approach (bottom)

## Tables

**Table 1** Overview of conducted PEF experiments and process conditions

Sample	E (kV cm <sup>-1</sup> )	W <sub>PEF</sub> (kWh kg <sub>DW</sub> <sup>-1</sup> )	Temperatu re (°C)
C- 25	0	0	25
P1 -25	20	0.55	25
P2 -25	20	1.11	25
C- 35	0	0	35
P1 -35	20	0.55	35
P2 -35	20	1.11	35
C- 45	0	0	45
P1	20	0.55	45

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-45

P2                    20                    1.11                    45

-45

C-                    0                    0                    55

55

P1                    20                    0.55                    55

-55

P2                    20                    1.11                    55

-55

C-                    0                    0                    65

65

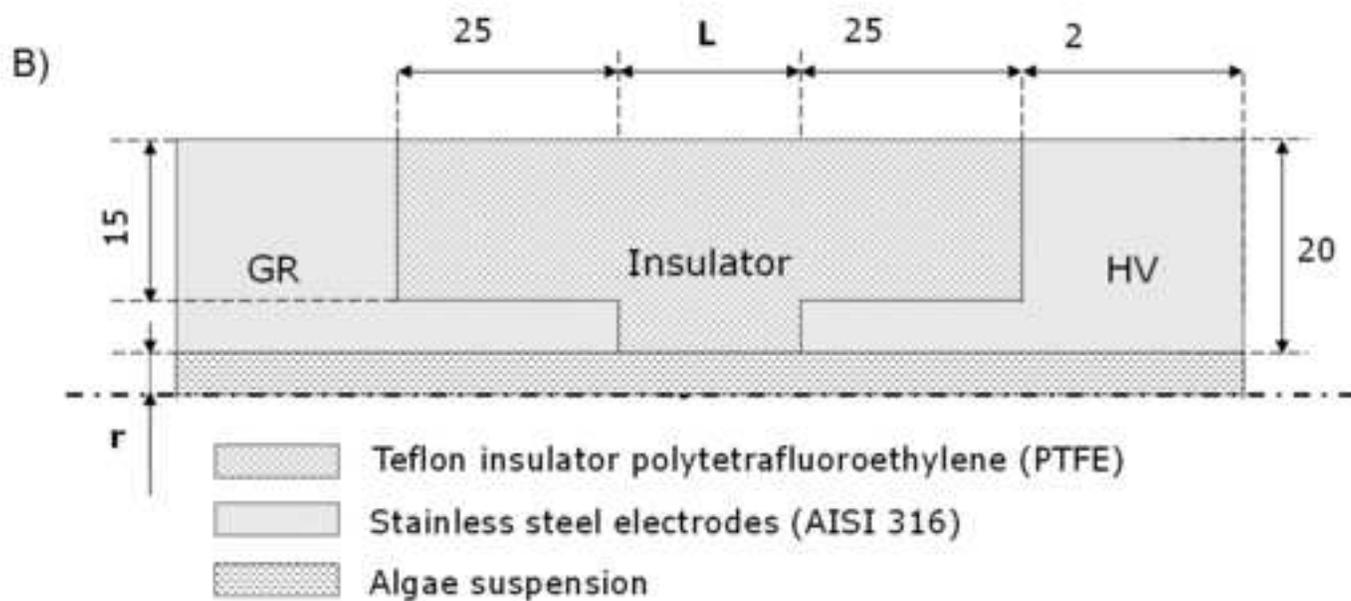
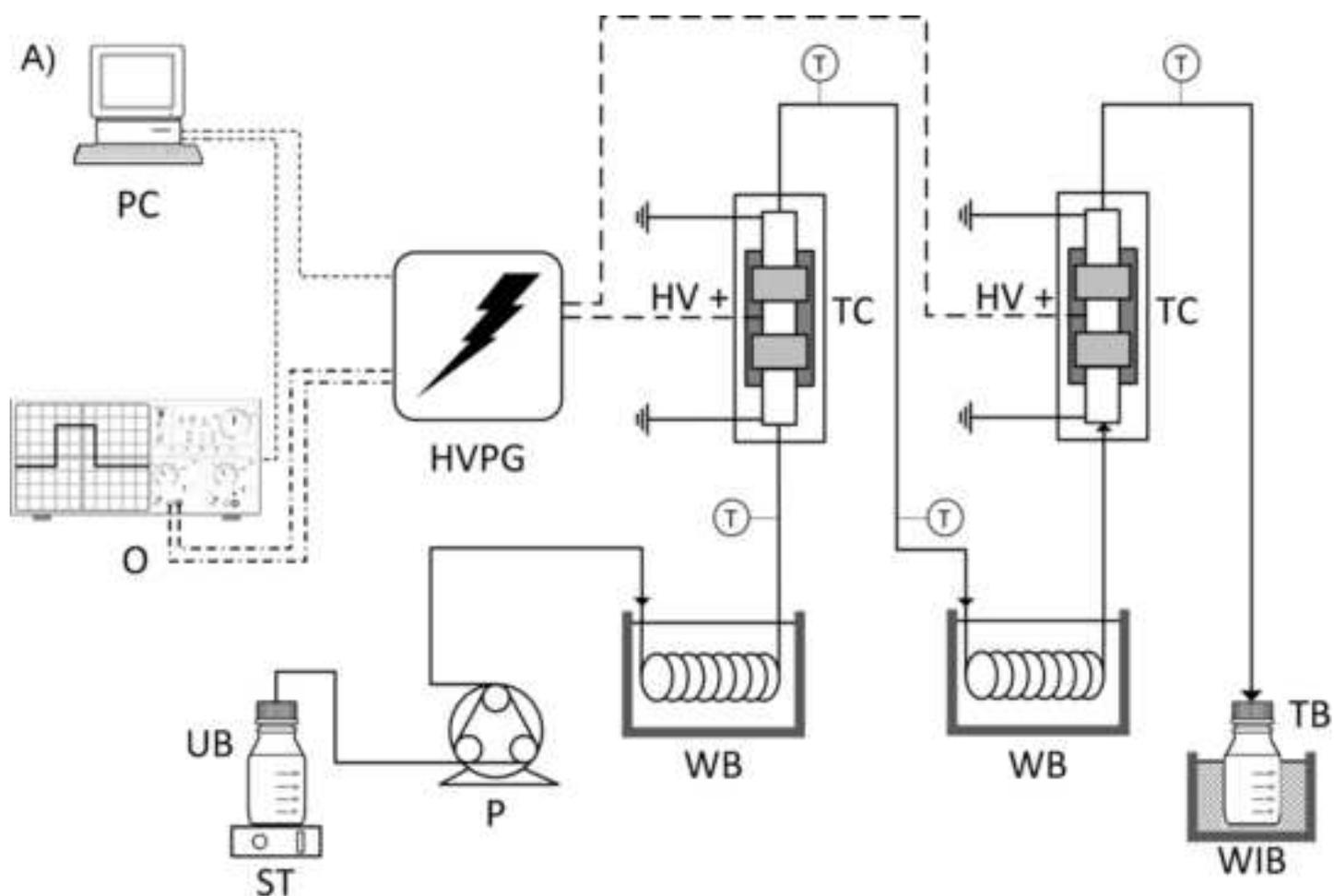
P1                    20                    0.55                    65

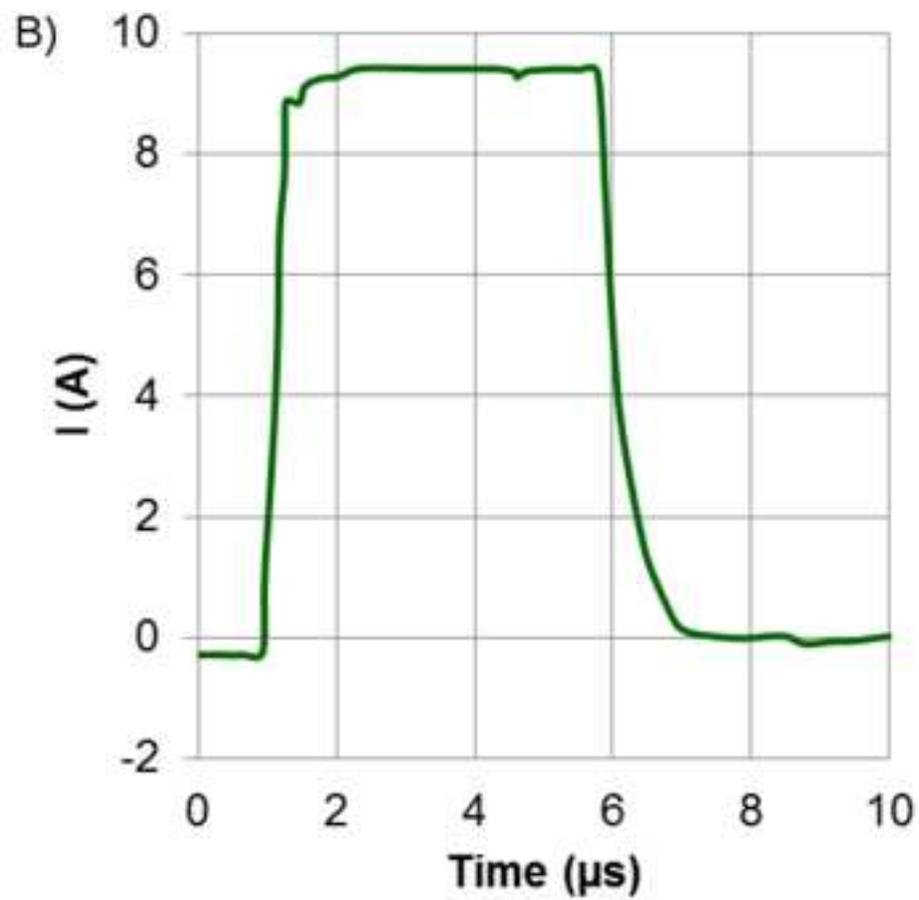
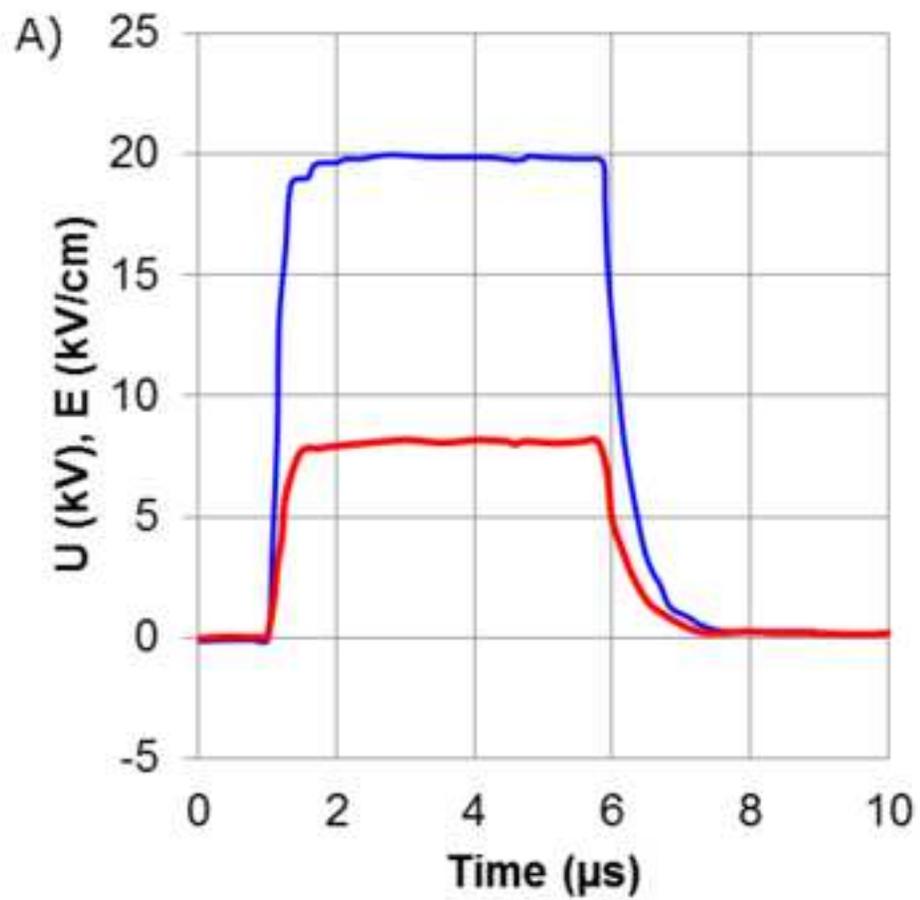
-65

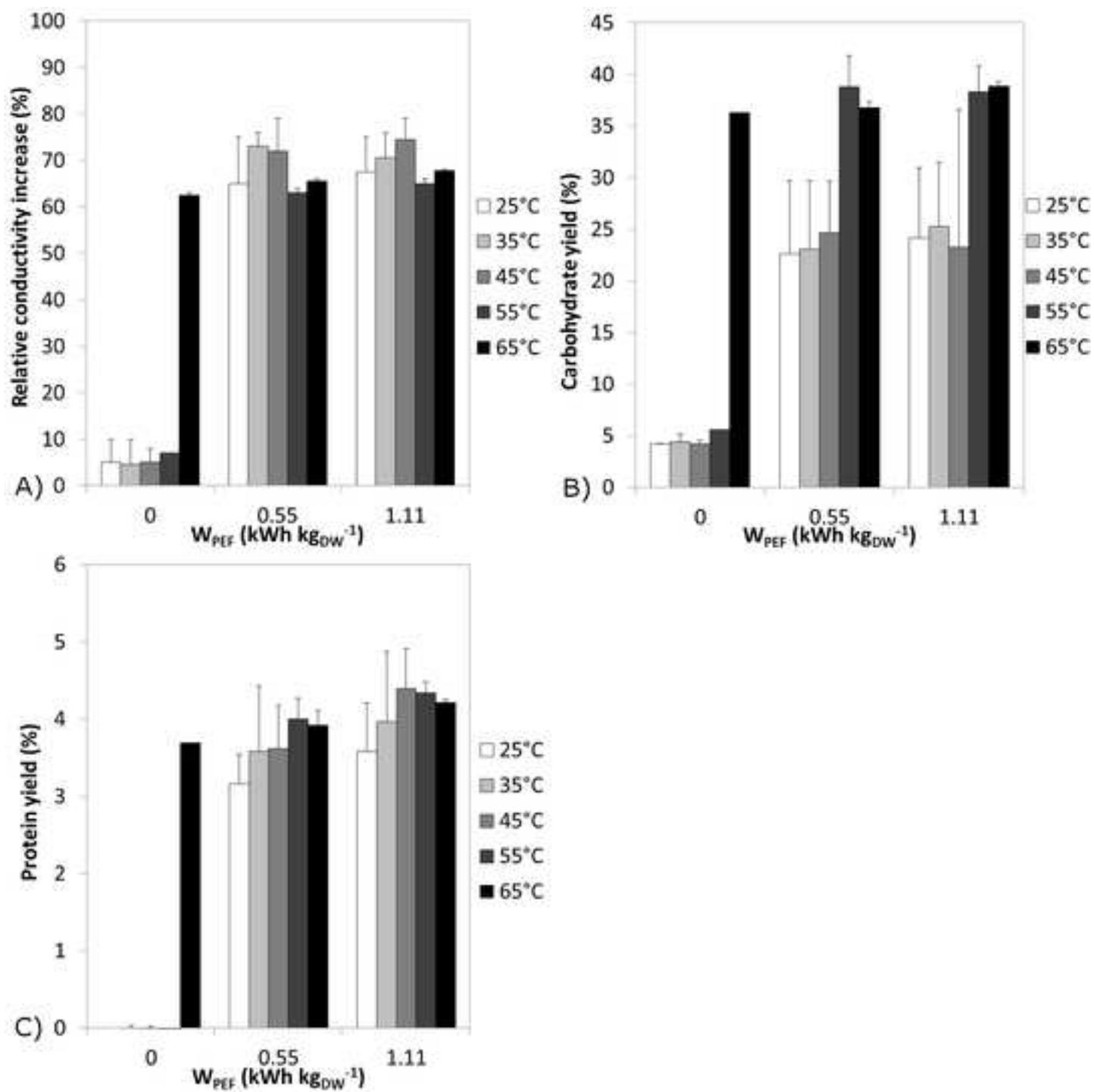
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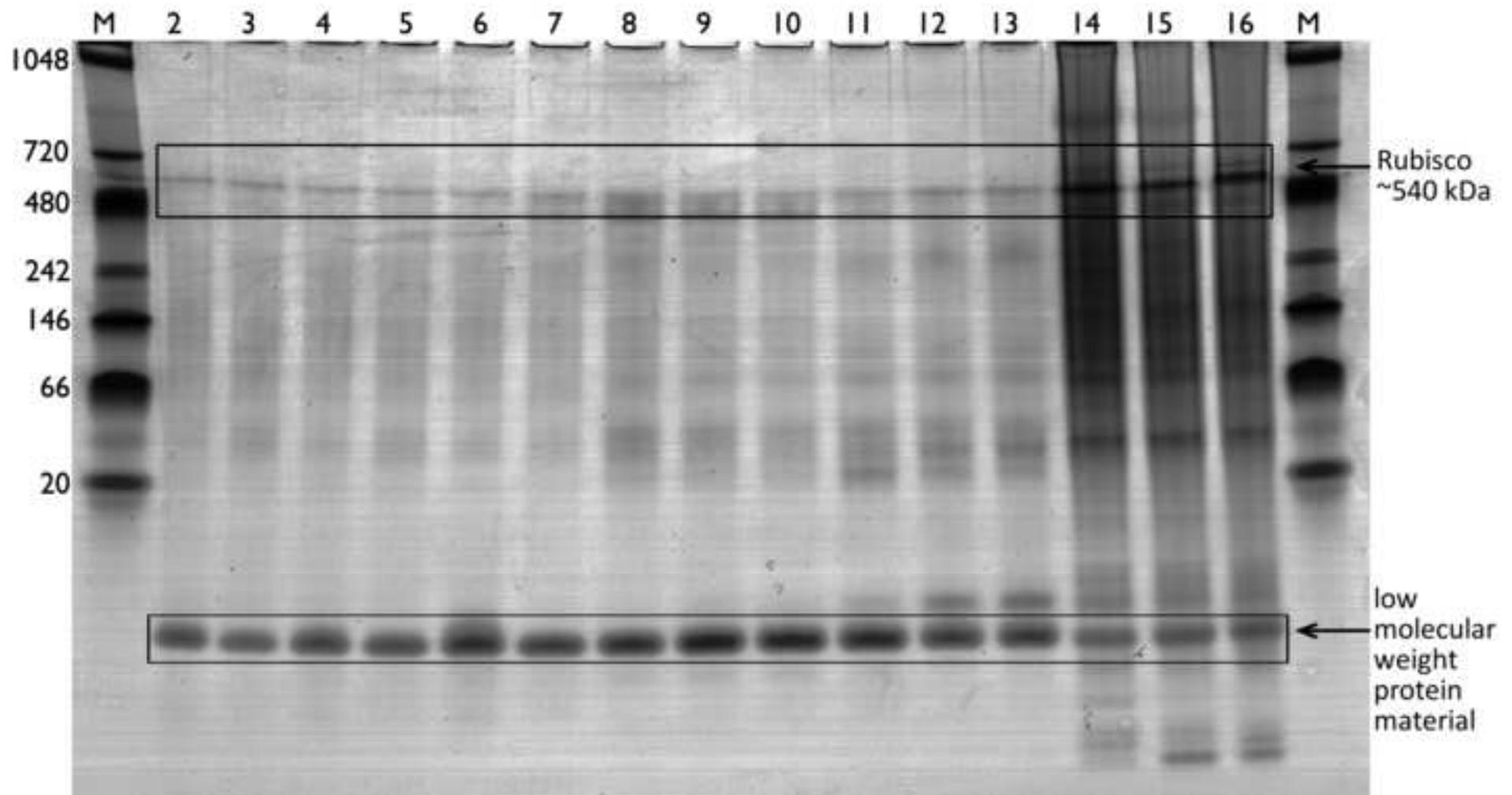
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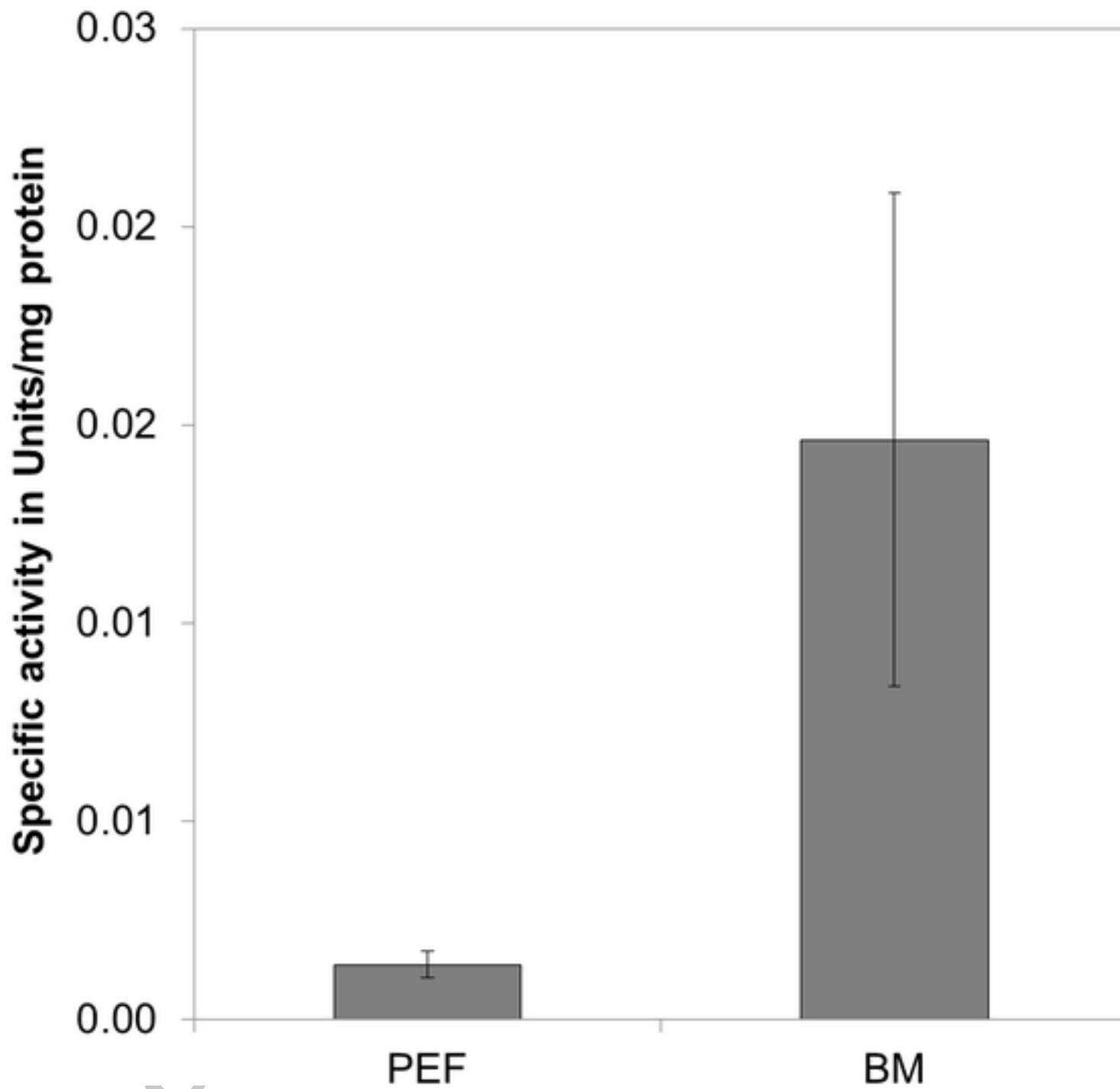
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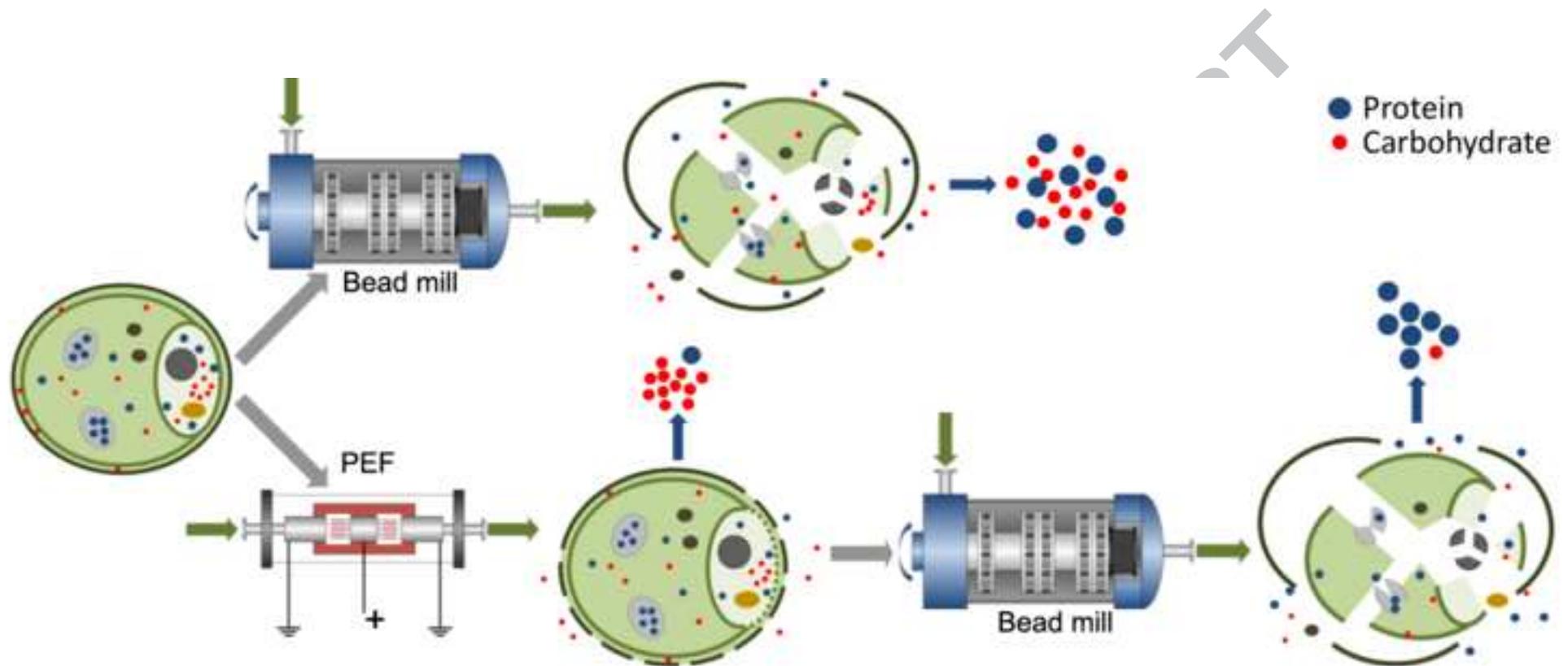


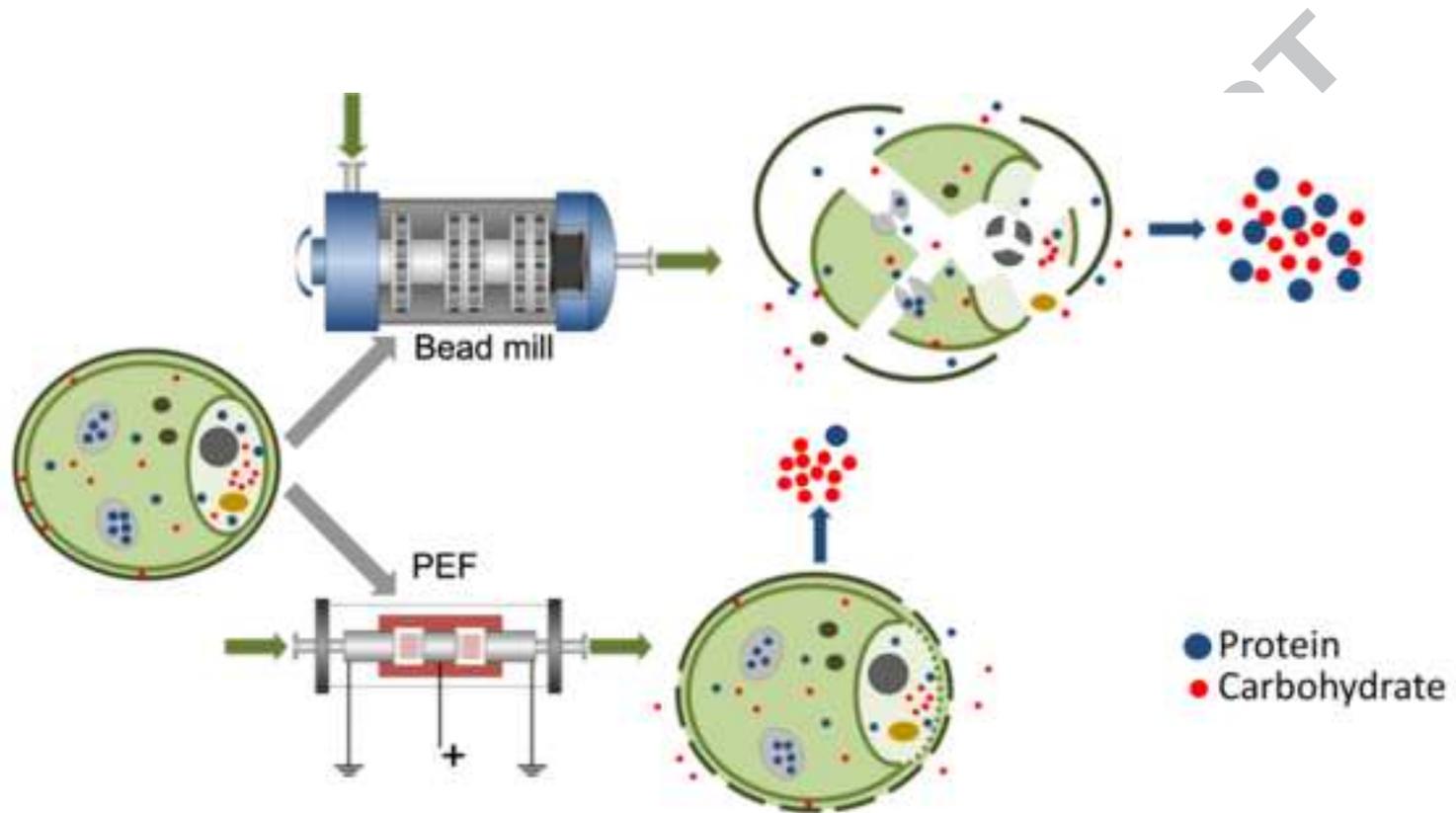












**Highlights**

- Algal cell permeabilization was achieved by applying pulsed electric field
- A synergistic effect was observed at 55°C for the release of carbohydrates by PEF
- PEF allows selective release of small water soluble components
- Over 95% of proteins are retained inside the microalgal cell after PEF

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